

LEYPUNSKIY, A.I., KAZACHKOVSKY, O.D., AFANAS'YEV, I.I., PINZHANSK,
M.V., KHAENYAROV, N.V., POYTO, M.S.

Uranium-cycled fast reactors. Atom. energ. 17 no.5 p45-548
N.Y., 1964.
(MRA 17 1c)

L 20049-65 EPF(c)/EPF(n)-2/EPR/EPA(s)-2/ENT(m)/EPA(bb)-2/EWP(b)/EWP(t)

Pr-4/Pb-4/Pt-10/Pu-4 SSD/AFWL/IJP(c) DH/WW/JD/JG

ACCESSION NR: AP4049534

S/0089/64/017/005/0345/0348

AUTHORS: Levpunskiy, A. I.; Kazachkovskiy, O. D.; Afrikantov, I. I.;
Rinkhasik, M. S.; Krasnoyarov, N. V.; Poydo, M. S.

TITLE: Sodium cooled fast reactors /9

SOURCE: Atomnaya energiya, v. 17, no. 5, 1964, 345-348

TOPIC TAGS: power reactor, liquid metal cooled reactor, fast
reactor/BN-350

ABSTRACT: The first fast-neutron power reactor now being designed in the SSSR (BN-350) is described. It is rated 1000 MW thermal and 350 MW electrical. Sodium coolant at 300C (total volume 165 m³) is heated in the reactor to 500C by about 200 fuel elements. The volume of the active zone (~2000 liters) and the power ratio (500 kW/liter) ensure a minimum use of fuel in the cycle. The ratio of the diameter of the active zone to its height (D/H) is 1.4 (D = 1.5

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L 20049-65

ACCESSION NR: AP4049534

(m , $H = 1.06 m$), the maximum sodium speed is 10 m/sec, the thickness of the breeder zones on the periphery and on the end is 60 cm. The construction permits the active zone size to be varied and to use different types of fuel elements. A ceramic fuel element is used consisting of a mixture of plutonium dioxide (19% Pu) and U²³⁸. Enriched (23%) uranium dioxide can also be used. The fuel rod is a stainless steel tube 5 mm in diameter and 0.4 mm thick, filled with pellets of the sintered fuel. The arrangement of the active and breeder zones is such as to produce a conversion ratio ~1.5. The internal conversion ratio is 0.62. The change in reactivity is 0.6% per month and is compensated by motion of central fuel elements with a reactivity margin of 1.4%, permitting 2 months' continuous operation. The shielding, control, and safety precautions are described. The sodium flows through a heat exchanger in which steam is produced at 430C and 50 atm pressure. Some improvements are suggested for future designs on the basis of the experience already gained in the design of the BN-350. Orig. art. has: 1 figure.

Card 2/3

L 11164-65 ENT(m)/EFF(c)/EFF(n)-2/EPR Pr-4/Ps-4/Pu-4 AEDC(b)/AFWL/BSD/SSD
DM

ACCESSION NR: AP4036523

8/0089/64/016/005/0407/0413

AUTHOR: Bagdasarov, Yu.Ye.; Kazachkovskiy, O.D.; Pinkhasik, M.S.; Py*shin, V.K.

TITLE: Study of unsteady operating conditions for natural circulation in multi-loop designs of nuclear reactors 19

SOURCE: Atomnaya energiya, v. 16, no. 5, 1964, 407-413

TOPIC TAGS: nuclear reactor cooling, liquid metal cooling, unsteady reactor operating condition, reactor emergency shutdown

ABSTRACT: The authors have developed a method of computation of unsteady operating conditions (emergency shutdown) for natural circulation in multiloop designs of nuclear reactors. The essential point to be considered is the delay factor which depends on the heat exchange between the coolant and the stationary parts of the loop (pipes); the geometry of the heat exchanger is important. Since exact solutions are impossible because of the complexity of the problem, the authors have worked out an approximate system of equations which was programmed on an electronic computer. In the computations, the delay, coolant flow rate, heat losses through insulation, and heat exchange between the steel structure and liquid sodium were considered. Experimental results are in good agreement

Card 1/2

L 14464-65

ACCESSION NR: AP4036523

with the computations. Of the factors mentioned above, the most important one is the heat exchange between the coolant (sodium) and the stationary part of the structure. The natural circulation is a reliable cooling method for nuclear reactors. Orig. art. has: 2 figures.

ASSOCIATION: none

SUBMITTED: 03Oct63

ENCL: 00

SUB CODE: NP

NO REF Sov: 003

OTHER: 000

Card 2/2

PINKHASIK, Z.

Whose interests does the public prosecutor defend? Okhr. truda i sets.
strakh. no.1:95 Jl '58.
(MIRA 11:12)

1.Tekhnicheskiy instruktor Belorusskogo respublikanskogo soveta prof.-
soyuzov, Minsk.

(Minsk--Housing)

ACCESSION #: AT4040806

S/3099/62/000/001/0189/0196

AUTHOR: Askarov, M. A.; Dzumerkas, N. D.; Pinkhasov, S. R.

TITLE: A study of the copolymerization of acrylonitrile with esters of acrylic acid

SOURCE: AN UzSSR. Institut khimii polimerov. Fizika i khimiya prirodnykh i sinteticheskikh polimerov, no. 1, 1962, 189-196

TOP C TAGS: copolymerization, acrylic ester copolymer, acrylonitrile copolymer, polymer structure, polymer physical property, propylacrylate, butylacrylate, amylacrylate, polymer solubility

ABSTRACT: The authors first describe the synthesis of n-propyl, n-butyl and n-amyl acr., ate by the simultaneous saponification and esterification of acrylonitrile in the presence of the appropriate alcohols, H_2SO_4 , and hydroquinone. After purification of both the ester and the acrylonitrile, their block copolymerization was then studied at 60°C. Measured amounts of the monomers were placed into ampules with a benzoyl peroxide catalyst (0.5% by weight), sealed and placed into an oven at 60°C for 32 hours. The yields were 63-91% of the theoretical. A detailed investigation of the properties of the copolymers at ratios of acrylonitrile to esters of 90:10, 75:25, 50:50, 25:75, 10:90, and 0:100 showed a consistent relationship between
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ACCESSION NR: A74040806

polymer structure, properties and the ratio of the monomers. Thus, no matter which ester was used, solid yellow copolymers with high values of specific viscosity and % N but limited solubility (only in dimethylformamide) were obtained at acrylonitrile: ester ratios of 90:10 and 75:25. At ratios of 50:50 and 25:75, soft yellow polymers were obtained with lower viscosity and % N but wider solubility, and at a 10:90 ratio, a transparent viscous polymer was obtained which resembled that from the pure acrylic esters (low specific viscosity and solubility in all organic solvents tested). Orig. art. has: 4 tables and 3 chemical equations.

ASSOCIATION: Institut khimii polimerov AN UzSSR (Institute of Polymer Chemistry,
AN UzSSR)

ENCL: 00

SUBMITTED: 00

OTHER: 010

SUB CODE: OC

NO REF Sov: 000

Card 2/2

deceased

L 25639-65 EPF(c)/EPR/EPA(s)-2/EWP(j)/EWT(m)/T Pe-4/Pr-4/Ps-4/Pt-10

RPL RM/WW/JT

ACCESSION NR: AP5005263

S/0291/64/000/006/0045/0050

49

39

B

AUTHOR: Askarov, M. A.; Pinkhasov, S. R. (Deceased)

TITLE: The synthesis and polymerization of some chloroacrylates

SOURCE: Uzbekschiy khimicheskiy zhurnal, no. 6, 1964, 45-50

TOPIC TAGS: plastic, acrylate, methacrylate, chloroalkyl acrylate, chloroalkyl methacrylate, polymer solubility, organic solvent, flame resistant plastic, corrosion resistant plastic

ABSTRACT: β -chloroethyl and δ -chlorobutyl acrylates and methacrylates were obtained in high yields by a direct esterification of ethylene or tetramethylene chlorohydrin with the corresponding acid in the presence of 3.5-4% H_2SO_4 added as catalyst. The obtained monomers polymerized readily under the effect of benzoyl peroxide, day- or UV-light, or heating to 60 \pm K. Glassy, colorless, transparent polymers were obtained which were insoluble in the organic solvents and only swelled in polar solvents (slightly in aromatic hydrocarbons). It was found that oxygen inhibits this polymerization. The insolubility was explained by crosslinking of double bonds formed as the result of a partial splitting off of HCl. The chloro-

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L 25639-65

ACCESSION NR: AP5005263

acrylates or chloromethacrylates obtained are prospective polymers which can be modified by the introduction of new groups and can be used for manufacturing flame-
or corrosion-resistant plastics. Orig. art. has: 3 tables and 1 formula. [BN]

ASSOCIATION: Nauchno-issledovatel'skiy institut khimii i tekhnologii khlopkovoy
tsellyulazy Goskomiteata khimicheskoy promyshlennosti pri Gosplane SSSR (Scientific
Research Institute of Chemistry and the Technology of Cotton Cellulose of the State
Committee of the Chemical Industry of Gosplan, SSSR)

SUBMITTED: 19Dec63

ENCL: 00

SUB CODE: OC, GC

NO REF Sov: 004

OTHER: 012

ATD PRESS: 3185

Card 2/2

TRUB, Israill' Ayzikovich; MONOKHOVICH, Eduard Isaakovich; MIKHAYLOVA, Ye.N.,
redaktor; PIMKHASOV, Ya.B., tekhnicheskij redaktor

[Using waste industrial heat in greenhouses and hothouses] Ispol'-
zovanie otbrosnogo tepla promyshlennosti v teplitsakh i parnikakh.
Tashkent, Gos. izd-vo Uzbekskoi SSR, 1955. 91 p. (MLRA 9:10)
(Greenhouses) (Heat engineering)

KUNYAVSKIY, M.P., kandidat ekonomicheskikh nauk, redaktor; TATUR, P.K.,
kandidat tekhnicheskikh nauk, redaktor; BONDARENKO, M.N., redaktor;
PINKHASOV, Ya.B. tekhnicheskiy redaktor

[Manual for machine-tractor station engineers and organizers of land
use] Spravochnik inzhenera-zemlestroytela MTS. Pod obshchei red.
M.P.Kuniavskogo i P.K.Tatur. Tashkent, Gos. izd-vo UzSSR, 1955. 342 p.
(MLRA 9:8)

1. Uzbek S.S.R. Ministerstvo sel'skogo khozyaystva. Upravleniye
zemlestroystva.

(Agricultural engineering)

MAKARKOVICH, Yu.A., ISHCHENKO, I.V., MIRONOV, T. A., TIKHONOV, V. N.

Bacteriological and immunological data on the significance of
autoimmune mechanisms in coronary and atherosclerotic diseases.
Zhur. mikrobiol., epid. i imun. 43 no. 123-37 Jan 1966
(vysl. 1961)

1. Translation contract number: QM 100. Submitted
January 1968.

PINKHALOVI N. .

Chem. Tech. Sci.

Dissertation: "Structure of the thread : needles o. knitting, actines."
15 Sep 47

Moscow Textile Inst.

SO Vecheryaya Moskva
Sum 71

PINKHENSON, D. M.

PL 175T72

USSR/Oceanography - Northern Sea Jul/Aug 50
Route

"Historic Stages in the Conquering of the Northern
Sea Route," D. M. Pinkhenson

"Iz v-s Geograf Obshch" Vol LXXXII, No 4,
pp 396-411

History of development of the northern sea
route from 15th century to now. Last figures
given are for 1936, when 14 ships navigated en-
tire route. Mentions industrial centers of
Kirovsk, Monchegorsk, Vorkuta, Anderma, Noril'sk,
Igarka, Anadyr', and Magadan. No data given on
postwar exploitation of route.

FID

175T72

PINKHENSON, Dmitriy Moiseyevich, kand. geogr. nauk, dots.; GAKKEL', Ya.Ya., doktor geogr. nauk, prof., red.; CHERNENKO, M.B., red.; FRISHMAN, Z.S., red.izd-va; KOLYAKOVA, O.I., tekhn. red.

[History of the discovery and adoption of the Northeast Passage] Istoriia otkrytiia i osvoeniiia Severnogo morskogo puti. Leningrad, Izd-vo "Morskoi transport." Vol.2. [Northeast Passage in the period of capitalism] Problema Severnogo morskogo puti v epokhu kapitalizma. Pod red. IA.IA.Gakkelia, M.B. Chernenko. 1962. 765 p.

(MIRA 17:3)

1. Leningrad. Arkticheskiy nauchno-issledovatel'skiy institut.
2. Deystvitel'nyy chlen Geograficheskogo obshchestva SSSR (for Chernenko)

PINKHUSOVICH, L. L.

7

107-4E2c

Problems of Metallurgy. Academy of Sciences of the U.S.S.R., Moscow, 1953. Mechanical Properties of Bessemer Low-Alloy Structural Steel. N. P. Shcherapov and A. G. Vodovozov. (409-411) (In Russian). An account is given of a comprehensive investigation of the mechanical properties of a low-alloy Bessemer steel (0.06-0.13% C, 0.28-0.64% Mn, trace-0.44% Si, 0.044-0.051% P, 0.028-0.048% S, 0.00-0.90% Cr, 0.00-0.58% Ni, 0.04-0.44% Cu, 0.013-0.018% N, 0.0008-0.00019% Hg, 0.00178-0.00165% O₂). The main conclusions drawn are: through alloying, the steel is actually less liable to brittle fracture than the corresponding O.I.L steel and has a higher yield point; the ageing properties and sensitivity to stress-concentration in cyclic loading remain relatively poor. It is suggested that by using other measures in addition to alloying, the properties of Bessemer steel can be improved still further. Investigation of Processes Occurring during the Tempering of Hardened Steel. K. F. Starodubov. (442-450). Changes occurring in hardened steel during tempering, mainly at 300-550° C, are described and explanations are proposed. Primary Structure of the Ingots and its Effect on the Properties of Steel. A. P. Pronov. (481-488). Unlike the finely crystalline primary structure of a carbon steel ingot, a dendritic one is characteristic of correct production conditions and results in good mechanical properties both at high and low temperatures. Factors governing the type of primary structure formed have been partially elucidated. Main Questions in the Rail Problem. L. L. Pinkhusovich. (407-408). The problem of rail quality as it has been dealt with in the U.S.S.R. is reviewed and the main factors involved are discussed.

AM
RG
JN
LW

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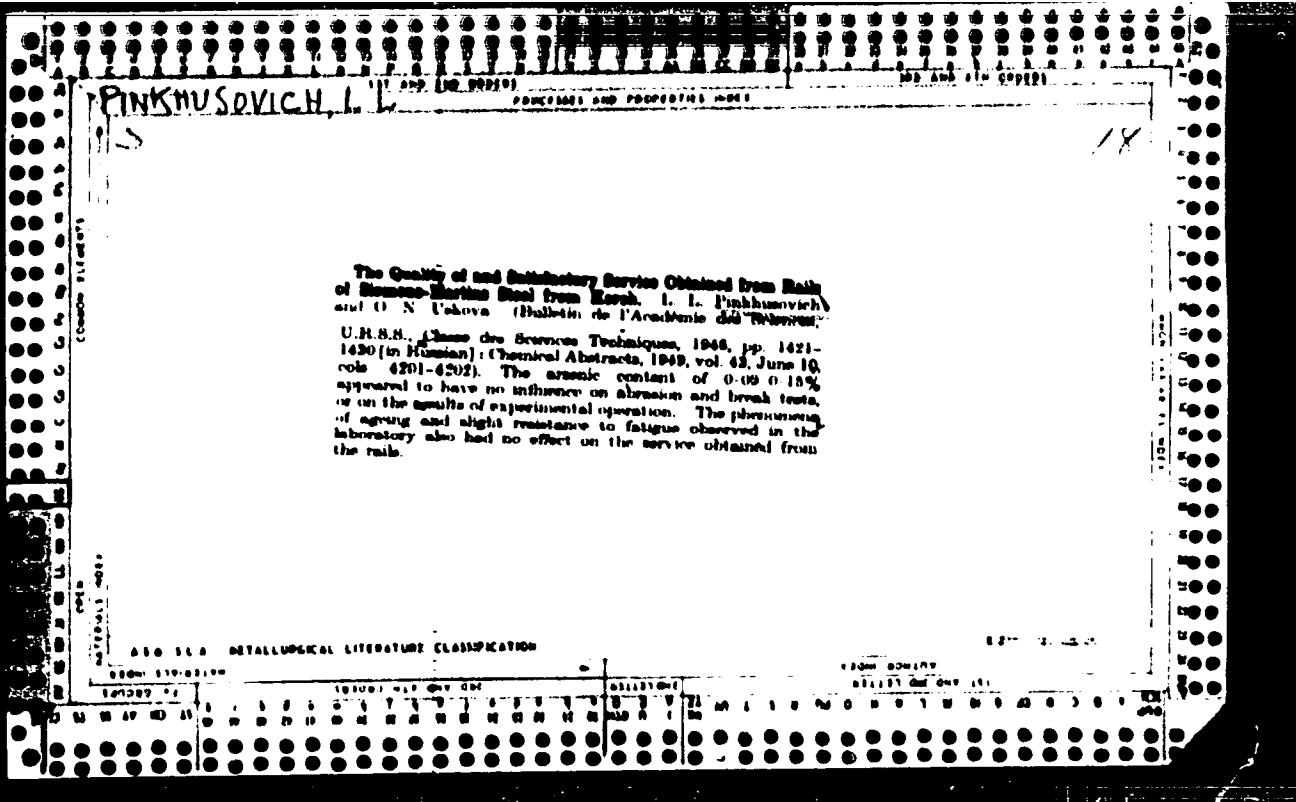
GRDINA, Yu. V.; BABICH, G. F.; GRUZDEV, D. S.; PINKHUSOVICH, L. L.

"Study of Kerch Arsenic Rails (Kerchenskikh mysh'yakovistykh rel'sov)," Iz. Ak. Nauk
SSSR, Otdel. Tekh. Nauk, No. 2, 1941. Submitted 30 Sep 1940.

Report U-1530, 25 Oct 1951

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5"



POPOVA, V.M., kandidat sel'skokhozyaystvennykh nauk, redaktor; BONDARENKO, M.N., redaktor; PINKHASOV, Ya.B., tekhnicheskiy redaktor

[Principles of livestock raising] Osnovy zhivotnovodstva. Tashkent, Gos. izd-vo Uzbekskoi SSR, 1955. 234 p. (MLRA 9:12)
(Stock and stockbreeding)

POPOV, Viktorin; LEVITINA, S.A., red.; PINKHASOV, Ya.V.

[Chardzhou-Kungrad; along the track with a notebook]
Chardzhou - Kungrad; s bloknotom po trasse. Tashkent,
Gos.izd-vo UzSSR, 1947. 29 p. (MIRA 16:8)
(Soviet Central Asia--Railroads--Location)

URUTSKAYA; VISHNIKOVA; BORISOV; PINKHASOVICH; MURADOV; RIGEL'MAN; OSERSKIY;
PIATOV; BOKSHEMAN; GORPISHCHENKO; YEREMENKO; ZHARKOV; POPOV; ROMANOVA;
SIDORENKO; TODIRIN; TIMOVSEYEVA.

Dmitrii Sergeevich Pavlov; obituary. Gaz. prom. no.1:56 Ja '58.
(Pavlov, Dmitrii Sergeevich, 1904-1957) (MIRA 11:2)

PINZHENSON, D.M.; SUROVTSEV, N.S.

Deficiencies in the training of students in economic geography.
Geog. v shkole 22 no.1:42-44 Ja-F '59. (MIRA 12:4)
(Geography, Economic--Study and teaching)

12-1-12/26

AUTHOR: Pinkhenson, D.M.

TITLE: D.I. Mendeleyev and the Utilization of Russia's Far North
(D.I. Mendeleyev i osvoyeniye kraynego severa Rossii)

PERIODICAL: Izvestiya Vsesoyuznogo Geograficheskogo Obshchestva, 1958,
1, pp 69-75 (USSR)

ABSTRACT: The article deals with Dmitry Ivanovich Mendeleyev, a great
Russian nineteenth century scientist, who stressed the great
importance of developing the Northern and Arctic regions.
There are 11 Russian references.

AVAILABLE: Library of Congress

Card 1/1

KOVYAZIN, N.; PINKHENSEN, D.¹⁷⁾

"Ukrainian Soviet Socialist Republic." Ol.T.Dibrova. Reviewed by
N.Koviasin, D.Pinkhensen. Geog. v shkole 19 no.4:75 Jl-Ag '56.
(Ukraine--Economic conditions)(Dibrova,Ol.T.) (MLRA 9:10)

PINKHENSON, D. M.

"Role of natural resources in the development of productive capacities" by L. I. Gurvich[deceased]. Reviewed by D. M. Pinkhenson. Issv. Vses. geog. ob-va 94 no.6:532-533 N-D '62.
(MIRA 16:1)

(Economic history) (Natural resources)
(Gurvich, L. I.)

PINKHENSON, D.M., kand. geograficheskikh nauk

Makarov's project of an icebreaker for an Antarctic expedition.
Inform. biul. Sov. antark. eksp. no. 36:41-44 '62.

(MIRA 16:4)

1. Gosudarstvennyy pedagogicheskiy institut imeni V. G. Kortseva.
(Ice-breaking vessels)

PINKHENSON, D.M.

Pages from the history of the White Sea - Baltic Canal. set.
Sev. 3:207-211 '62. (MIRA 15:3)

1. Pedagogicheskiy institut imeni A.I.Gertseva, Leningrad.
(White Sea--Baltic Canal)

PINKHENSON

BARTSOV, Nikolay Nikolaevich, dotsent, kand.geograf.nauk; BONIFAT'YVA,
Lidiya Ivanovna, dotsent, kand.geograf.nauk; BUREVSKO, Sergey
Fedorovich, dotsent, kand.geograf.nauk; GITLITS, Semen Abramovich
vich, dotsent, kand.ekonom.nauk; GUREVICH, Prim Vladimirovich, prof.;
DARINSKIY, Anatoliy Viktorovich, dotsent, kand.geograf.nauk; DOLININ,
Aleksay Arkad'yevich, dotsent, kand.geograf.nauk; DOROSHKEVICH,
Lyudmila Ivanovna, dotsent, kand.geograf.nauk; YEFIMOVA, Yelena Se-
menovna, kand.geograf.nauk; LAVROV, Sergey Borisovich, dotsent, kand.
geograf.nauk; LEDOVSKIY, Stepan Ivanovich, dotsent, kand.geograf.
nauk; NEVEL'SHTEYN, Grigoriy Solomonovich, dotsent, kand.geograf.
nauk; NIKOLAYEVA, Nadezhda Vasil'yevna, dotsent, kand.geograf.nauk;
OGADEV, Vladimir Artem'yevich, kand.geograf.nauk; PINKHENSON,
Dmitriy Moiseyevich, dotsent, kand.geograf.nauk; POSPELOVA, Nata-
liya Georgiyevna, prof., doktor ekonom.nauk; SEMREVSKIY, Boris Nikola-
yevich, prof., doktor geograf.nauk; SUTYAGIN, Pavel Grigor'yevich,
dotsent, kand.geograf.nauk; SHTEYN, Viktor Moritso维奇, prof., doktor
ekonom.nauk; YEROFEEV, I.A., red.; SHIROKOVA, N.P., red.; TYUTYUNNIK,
S.G., red.kart; BORISKINA, V.I., red.kart; KOZLOVSKAYA, M.D.,
tehn.red.

[Economic geography of foreign countries; student manual] Ekonomi-
cheskaya geografija zarubezhnykh stran; posobie dlja studentov. Moskva,
Gos.ucenno-pedagog.izd-vo M-va prosv.RSFSR, 1960. 702 p. 4 maps

(Geography, Economic)

(MIRA 13:12)

PINKHORN, D.M.

Terrorists "organizers" and interests in developing the North.
Inv.Vses.geog.ob-va 88 no.4:382-384 J1-Ag '56. (MLRA 9:10)

(Russia, Northern--Geography, Economic)

GEMBEL', Aleksandr Vasil'yevich; PINKHENSON, D.M.; PODOPLELOV, N.Ya.

[Natural resources of the U.S.S.R. serve the building of communism] Prirodnye bogatstva SSSR na sluzhbe kommunisticheskogo stroitel'stva. Leningrad, Ob-vo po rasprostraneniu polit. i nauchn. znanii RSFSR, 1959. 33 p. (MIRA 15:9)
(Natural resources)

PINKHENSON, Dmitriy Moiseyevich, doktor geogr. nauk; GORDIYENK^O.
P.A., kand. geogr. nauk, nauchn. red.

[New developments on the map of the U.S.S.R.; basic changes
in the geography of the country during the Soviet period]
Novoe na karte SSSR; osnovnye izmenenija v geografii strany
za sovetskiy period. Leningrad, Ob-vo "Znanie" RSFSR, 1964.
43 p. (MIRA 18:3)

AL'TMAN, L.P.; PINKHENSON, D.M.; CHERTOV, L.G.; SEMEVSKIY, B.N., prof.,
nauchnyy red.; VOROB'YEV, G.S., red.izd-va; GURDZHIYEVA, A.M.,
tekhn.red.

[Geography of great works] Geografiia velikikh rabot. Leningrad,
Vses. ob-vo po rasprostraneniiu polit. i nauchn. znanii. Leningr.
otd-nie, 1960. 39 p. (MIRA 13:?)
(Russia--Industries)

LINDENBATH, C. J., LINDNER, F. H., LINDNER, H. A. (Monks).

and pharmacological properties of the extract in view of the greater biological activity. *J. Physiol.*, 1941, 40, 193. M.R.C.P. (Lond.).

PINKHOSEVICH, Ye. G.

X-ray archives in a therapeutic institution of the dispensary type. Vest. rent. i rad. 40 no.2:52-54 Mr-Ap '65.

(MIRA 18:6)

1. Kafedra rentgenologii i radiologii (zav.- prof. L.D. Linderbraten) i Moskovskogo ordena Lenina meditsinskogo instituta imeni Sechenova i rentgenovskoye otdeleniye (zav. V.I. Palshkova) polikliniki No.1, Moskva.

PINKHUSOVICH, I. I.

124-11-13556

Translation from: Referativnyy Zhurnal. Mekhanika, 1957, Nr. 11, p. 112 (USSR)

AUTHOR: Pinkhusovich, I. I.

TITLE: On the Notch Sensitivity of Steel Railings
(O chuvstvitevnosti rul'sovoy stali k nadrezam)

PERIODICAL: V sb.: Prochnost' metallov., Moscow, AN SSSR, 1956, pp 112-115

ABSTRACT: Bibliographic entry.

Card 1 1

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5

PINKHUSHOVICH, L. E., + ed.

Steel rails made from the ore of the Verkhnera river basin, Novaya, 17 km. from Arkhangelsk, Russia.
1948. Length - 1570.

15000

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CIA-RDP86-00513R001340920007-5

~~U.S. Embassy
In Russia;
Academy of Sciences, Moscow
In case of~~

cc: [redacted]

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5"

MOGUCHIY, Leonid Nikolayevich; SAVITSKIY, Ye. N., otvetstvennyy redaktor;
RUMYUSOVICH, L. L., redaktor izdatel'stva; ZEMLYAKOVA, T. I.,
tekhnicheskiy redaktor.

[Working metals under pressure] Obrabotka metallov, davleniem.
Moskva, Izd-vo Akad.nauk SSSR. 1957. 198 p. (MLRA 10:4)
(Metalwork)

BARDIN, I.P.; NIKONOV, A.G.; PINKHUSOVICH, L.L.

Investigation of the hardenability of wheel steel subjected to
sliding friction. Trudy Inst.met.AN SSSR no.1:114-119 '57.
(MIRA 10:11)
(Steel--Testing) (Car wheels)

NIKONOV, A.G.; PINKHUSOVICH, L.L.

Investigating trial batches of low-alloy wheel-grade steel. Trudy
Inst. met. no.11:90-98 '62. (MIRA 16:5)
(Chromium-vanadium steel--Testing)

NIKONOV, A.G.; PINKHUSOVICH, L.L.

Wheel bandage wear. Trudy Inst. met. no.4:250-253 '60.

(Car wheels--Testing)
(Mechanical wear)

(MIRA 14:5)

MINKHUSOVICH, L.L.; NIKONOV, A.G.

Effect of hardening methods on the wear resistance and fatigue
crumbling of all-rolled wheels. Izv. vys. ucheb. miv.; chern.
met. no.10:126-131 '60. (MIRA 13:11)

1. Institut metallurgii im.A.A.Baykova AN SSSR.
(Steel--Fatigue) (Wheels--Testing)

S.45 8-12-71 :2 :
A.61.463

AUTHORS: Pinkhusovich, L.L.; Nikonov, A.G.

TITLE: The Effect of the Quenching Method on Wear Resistance and on the Fatigue Crumbling of Rolled Seamless Wheels

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Chernaya metallurgiya, 1960, No. 10, pp. 126 - 131

TEXT: Rolled RR wheels are quenched in the USSR in a horizontal position by water jets directed on the rolling surface from tangentially placed pipes. This method has been compared with the U.S. method of quenching wheels in vertical position by rotation with a wheel rim portion submerged in water (Refs. 5 - 8). The test wheels were of standard steel with 0.53% C; 0.69% Mn; 0.27% Si; 0.02% P and 0.02% S. The hardness and the mechanical properties of wheels hardened by the U.S. method were better, they had pearlitic structure through the work portion and higher wear resistance in comparative tests on a IA Amsler test machine. In fatigue crumbling tests on the same Amolor machine, specimens taken at 12 and 30 mm distance from the surface of the wheels treated the U.S. way withstood 690,000 and 875,000 cycles, compared with 630,000 and 660,000 cycles withstood by

Card 1/2

S-145/60 1960-10-12 :
A161/A03C

The Effect of the Quenching Method on Wear Resistance and the Fatigue Crumbling of Rolled Seamless Wheels

specimens from same spots treated in the Soviet-way. The higher crumbing resistance is explained by the different metal structure having a higher resistance to cracks formation. The conclusion was made that wheels quenched in a vertical position by periodical submersion of a portion of the rim into a running water will have higher resistance to wear and fatigue crumbing. There are 5 figures, / 2 tables and 4 references: 5 Soviet and 3 English.

ASSOCIATION: Institut metallurgii im. A.A. Baykova AN SSSR (Institute of Metallurgy im. A.A. Baykov of the Academy of Sciences USSR)

SUBMITTED: November 26, 1954

Card 2/2

PINKHUSOVICH L.L.

PURPOSE: This book is intended for students and engineers.

SCOPE: The article in this collection present historical data on the development of some of the basic methods of bond formation and disconnection in theory and practice, especially on the thermodynamic approach. Many of the topics have been developed in other articles or books and some of the present state-of-the-art information can be found there. The emphasis is placed on the future, attempting to indicate what may be expected in the future. Attention will be given to some of the more advanced topics, such as the use of molecular beam techniques, for further refinement, and some of the more recent developments in the field.

16
Savchenko, I. S., Corresponding Member, USSR Academy of Sciences, Professor, Director of Technical Sciences, Institute of Metalurgy, and L. A. Rayev, USSR Academy of Sciences, Scientific Study of the Rolling Process

This article is an extensive survey of scientific writing on the rolling process published in various countries between the two world wars. The article deals with historical development, principle, theoretical and practical problems of rolling, rolling equipment, rolling mill and strip, force and power relations, deformation, ringed rolling, and empirical methods of rolling.

Historical Information on the Development of Standardized Standards for the Interpretation of Results of a Test on the Basis of a Mathematical Procedure. However, and since procedures to determine the weight and type of load, instruments used, conditions of the test, and conditions of the material tested are not mentioned, use of these standards is limited to the interpretation of other results obtained by other methods. The following table summarizes some of the available information on the development of standards for further legitimate use.

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5"

PUBLICATIONS, ETC.

16(e) PAGE 2 BOOK REVIEWS 807/778
 Available and 8000. Institute metallurgists
 Sovremennye problemy metallicheskogo (Modern Problems in Metallurgy)
 Moscow, Izdatelstvo Akademii Nauk, 1959. 640 p., 5,000 copies printed.
 Ed. by A.M. Semenov, Corresponding Member, USSR Academy of
 Sciences, Head of Publishing House, V.G. Krasnolutskiy, and
 A.I. Baranov, Tech. Ed. of V.T. Polyakov.
 NOTE: This book is intended for scientific and technical per-
 sonnel. This is a collection of articles on certain aspects of
 modern metallurgy. The book is dedicated to Academician
 Ivan Pavlovich Martyn on the occasion of his 75th birthday. The
 book is divided into seven parts. The first part consists of
 two articles presenting a brief account of the present state of
 professional activity of the Soviet metallurgist. It includes an
 article by Ivan Chukhrov, "The Present State of Soviet Metal-
 lurgy," describing their meeting with Martin in Moscow and also an
 article by the author of this review, "Metallurgy in the United States."
 The second part consists of three articles dealing with raw materials and
 their use in metallurgy. The third part deals with the major
 problems of the Soviet metallurgist. The fourth part consists of 25 articles dealing with
 various aspects of the metallurgy of pig iron and steel,
 copper, zinc, and aluminum. The fifth part concerns creating the metal-
 lurgical basis for the forming of various metals. The sixth part consists of three
 short articles discussing specific problems of metallurgy. The seventh part consists of three
 articles on the strength and toughness of materials. The last part deals with general problems of physical metal-
 lurgy. References are given after each article. No
 table of contents.

Books on metallurgy

Modern Problems in Metallurgy 807/778
 Compiled by I.I. Doctor of Chemical Sciences, Metallurgist
 Institute Izmail A.J. Baykov, As MSAI. Chemistry of
 Materials
 Semenov, K.P. [Candidate of Technical Sciences, Metallurgist]
 Gleiser, Tatyana of Petrov Metallurgy (Investigating the
 Strength and Toughness of Iron-Carbon Steels)
 By Basic Treatment
 Pashkevich, Ilya and Z.O. Prudnik. Investigating Fatigue
 Strength of Holes Containing Aneurysis
 Pashkevich, P.D., L.N. Krasnolutskiy, and V.A. Leont'ev (Metal-
 lurgical Processing Equipment), Increasing the Strength and
 Durability of Railways Rails by Oil Quenching
 600

GENERAL PROBLEMS IN METALLURGY
 By Semenov, K.P. [Candidate of Technical Sciences, Dispenses
 (the same) practice for the Design and Planning of Metallurgical Plants]
 Available: General Plans of Metallurgical Plants
 Date 12/13
 600/44
 6/15/59

AUTHORS: Ivanovskiy, V.P., V.V. Kuznetsov, A.A. Slobodchikov, A.I. Tikhonov, A.N. Vinogradov

TITLE: Fatigue strength of heat treated rails containing Arsenic (Estalochnaya pravilnost' termicheskoi obrabotki tinnikh rel'sov, soderzhaushikh arzen yak)

PERIODICAL: Stal', 1955, N 9, p. 904-905 (SSh)

ABSTRACT: The influence of the type of hardening process on rails containing arsenic on their fatigue strength was investigated. Experiments were carried out on rail specimens of the R-50 type, 1.5 m long containing various proportions of carbon (0.6 to 0.8%) and arsenic (0.12 to 0.23%) from four heats (table 1). During thermal treatment, specimens were charged into a cold electric furnace and heated to the martensite temperature (800 to 820°C) on average during 1 hour and then cooled to this temperature for 30 minutes. Two kinds of hardening were tested: 1) surface hardening of the head with water sprays in a machine flame-furnace and subsequently self tempering at 330 to 400°C; 2) hardening by immersion in oil with cooling to room temperature. Fatigue strength of hardened specimen was tested in a special machine, described in ref. 1 and 2.

Card 1/3

4.1.1.5.4-173
Fatigue Strength of heat treated rails for railway service.

The experimental results are given in table 2 and fig 1 to 7. It was found that the arsenic content up to 0.23% has no influence on the fatigue strength of rails which confirmed earlier findings (ref 1). The dependence of the fatigue strength of rails on the type of thermal treatment is shown in fig 1. Annealing in water increases the fatigue strength of rails by about 30% (fig 1). The results of the fatigue strength of treated rails showed a small scatter spread, which may be accounted for by the variation of the heat treatment. The results of the fatigue strength of untreated and decarburized rails are given in fig 2. The degree of decarburization (about 1%) did not influence the fatigue strength. Within the depth of the zone of pure ferrite (ca. 1/7). If it is assumed that the limiting conditions of thermal treatment are maintained, the present results can be used to determine the required heat

Card 2/3

Fatigue Strength of Heat Treated rails containing Arsenic
the fatigue strength of rails. There are 6 figures,
3 tables and 6 references of which 3 are Soviet and
3 English.

ASSOCIATION: Institut Metallurgii A. SSSR (The Institute of
Metallurgy of the Academy of Sciences of the USSR)

Card 3/3

SOV 137 57 11 22645

Translation from Referat na v zhurnal Metallurgiya 1957 Nr. 1 p. 24 SSSR

AUTHORS Bardan' P. Nekrasov A.G. Pyrkusovich I.I.

TITLE The Hardenability of Wheel Steel Due to Sliding Friction
(Ussledoaniye zakari azemost' stali pri tretye skel zhemnya)

PERIODICAL Tr. In-ta metallofiz. AN SSSR 1957 Nr 1 pp 114-119

ABSTRACT A new method is described for testing the resistance of steel against the formation of chipped-out hollows on the rolling surface of railway wheels (W) through braking action. The reproduction of the hardened layer (H1) on the surface prior to the formation of the chipped-out hollows was accomplished under laboratory conditions on a special machine on which the specimen (S) of steel tested was fixed in a stationary position while the local heating was accomplished by the friction of a rotating W 1000 mm in diam with the rim machined in the shape of the railhead profile. The hardness on the rolling surface of the W was $H_B = 300$, i.e., equal to the hardness of the rail. The length of the S tested was equal to the width of the rim of the W; their width was 6.0 and thickness 40 mm. S were carefully

Card 1 3

SOV 137 -7 . . 1264-

The Hardenability of Wheel Steel Due to Sliding Friction

ground on four sides, washed with gasoline and fitted into the grinding slot of a yoke which was made of a piece of W rim, and fastened with a wedge. This ensured heat elimination at about the same rate as in the rim of a full-scale W. Thereupon the rail-shaped W was set into rotation and, when the surface of W reached a constant linear speed (60 km/hour) the S was pressed to the W with a force regulated by the safety valve of the hydraulic transmission and kept in this position for a specified period of time. To decrease the sticking of metal onto the surface of the W during its slipping on the S it was wetted with machine oil. Upon the expiration of the time of holding the specimen under pressure, the valve on the hydraulic transmission was switched over and the S was moved away from the W. The experiments were carried out on S of carbon steel with 0.59% C holding at a constant pressure of 1250 kg during 3, 5, and 8 sec and at different loadings (750, 1000, 1250 and 1500 kg) for a constant duration of 5 sec. In addition S of 7 grades of wheel steel were tested. After the tests the structure was investigated, microhardness was measured over the section of the S, and the HL was analyzed by X-ray diffraction. The evaluation of the resistance of the W against the formation of chipped out hollows through the braking action was made according to the hardness and the thickness of the HL. Tests were also performed on the formation of HL at low temperature (-60°C) in order to ascertain the effect. Card 2-3

SOV 137 57 11 22643

The Hardenability of Wheel Steel Due to Sliding Friction

of the temperature of the S on the thickness and hardness of the HL and also the tendency towards the formation of cracks without the application of mechanical action. The results of the investigations are adduced and recommendations are made on the manufacture of integrally-welded W resistant to the formation of chipped out hollows through braking action.

I - G

Card 3 3

Translation from: Referativnyy zhurnal Metallurgiya i Tsvetnaya metalloobrabotka No. 11 p. 8 USSR

AUTHOR: Pinkhusovich, I. L.

TITLE: On the Notch Sensitivity of Rail Steel (O chuvstvitev k nadrezam relesovoy stali k nadrezam)

PERIODICAL: V sb.: Prochnost' metallov. Moscow, AN SSSR, 1951
pp 112-116

ABSTRACT: The rolling-load fatigue test method was employed to investigate the notch sensitivity (NS) of rails made of bessemer steel, steel containing As, and four grades of alloyed steel: Mn steel, Cr-Ti steel, Cr-V steel, and 3 percent Cr steel. The NS was evaluated from the area of "fatigue spots" at the rail's surface. It is established that rails made of bessemer and Mn steel, which were not subjected to heat treatment, exhibited a considerable degree of NS. Low-alloy steel rails, which had been heat treated, exhibit a lower degree of NS than rails which had not been heat treated. The lowest NS was observed in Mn and Cr-Ti steels after heat treatment, and the highest in Cr and Cr-Ti steels which had not been heat treated. Cr-V steel exhibited a high degree of NS. Rails made of zirconium-bearing steel

Card 1/2

On the Notch Sensitivity of Rail Steel

ing As exhibit a lower NS than those made of bessent or alloyed steel.

M. S.

1. Steel-Fatigue-Test results
2. Rails (Railway)-Mechanical properties-Test results

Card 2/2

137-1957-12-22894

Translation from: Referativnyy zhurnal, Metallurgiya, 1957, Nr 12, p 9 (USSR)

AUTHOR: Pinkhusovich, L. L.

TITLE: On the Development of the Metallurgical Industry in Kerch (Ob usloviyakh razvitiya kerchenskoy metallurgii)

PERIODICAL: Tr. Insta metallurgii AN SSSR, 1957, Nr 1, pp 245-248

ABSTRACT: The accomplishment of a fuller development of ferrous metallurgy in the USSR will necessitate a wider exploitation of the ores of the Kerch deposits. An expansion of the metal production, based on the exploitation of Kerch ores is economically most profitable under a compound exploitation of the ores, which, along with steel, would also yield phosphate fertilizers and V. A number of institutes have been instructed to investigate the mineralogical composition of all varieties of the Kerch ores, to perform a detailed analysis of the geology of the area and of the chemical composition of the exploited sections, to complete the geochemical mapping of deposits which have been explored, and to work out plans for terracing and the prevention of slides. A large amount of work is planned for the processes of concentration, agglomeration, and metallurgical reduction.

A Sh.

Card 1 1

1. Metallurgy-USSR 2. Geophysical surveying 3. Ores-Determination

BLINOV, G.I.; MAKAROV, I.A.; PINKHUSOVICH, R.L.

Using radioactive control and regulation devices in hydrogenation
plants. Khim. i tekhnicheskaya promst. i masel 4 no.1:15-19 Ja '59.
(MIRA 12:1)

(Radioisotopes--Industrial applications)
(Liquid level indicators)

HI TE 31, .

"*Observations on the life, manners, customs, &c., of the people.*"
p. 50. (1787) "They are very fond of cards, and play them every day."

27. Minerals and Vitamins in the Control of Diabetes Mellitus. This article discusses the role of minerals and vitamins in the control of diabetes mellitus. It states that the disease can be controlled by diet, exercise, and insulin. It also discusses the use of minerals and vitamins in the treatment of diabetes mellitus.

28. Minerals and Vitamins in the Control of Diabetes Mellitus. This article discusses the role of minerals and vitamins in the control of diabetes mellitus.

29. Minerals and Vitamins in the Control of Diabetes Mellitus. This article discusses the role of minerals and vitamins in the control of diabetes mellitus. It states that the disease can be controlled by diet, exercise, and insulin. It also discusses the use of minerals and vitamins in the treatment of diabetes mellitus. The article notes that the presence of sodium and potassium in the animal's diet, balance ratios of magnesium, calcium, sodium, and potassium, without any disturbance of the animal's diet, are definitely involved. Other discussers discuss diagnosis and treatment, and notes that preventive supplementation of diet with minerals and vitamins is the easiest and surest control method.

[to list of references]

1/1

TUSZKIEWICZ, Maria; PINKIEWICZ, Halina; JABLONSKI, Leon

Preliminary investigations on antigen relationship of atypical
bacilli cultured from man. Ann. Univ., Lublin sect.D 16:229-237
'61.

1. Z Katedry i Zakladu Mikrobiologii Lekarskiej Wydzialu Lekarskiego
Akademii Medycznej w Lublinie Kierownik: prof. dr Jozef Parnas.
(MYCOBACTERIUM) (ANTIGENS)

SZCZYGIELSKA, Jadwiga; PIKIEWICZ, H.; BIERNACKI, Marian; PLESZCZYNSKA, Ewa

Studies on the pathogenesis of experimental influenza in white mice.
Communication II. Med.dosw.mikrob. 12 no.1:15-19 '60.

1. Z Katedry Mikrobiologii Lekarskiej A.M. w Lublinie. Kierownik:
prof.dr J. Parnas.
(INFLUENZA exper.)

ŚZCZYGIELSKA, J.; PLESZCZYNSKA, E.; BIERNACKI, M.; PARNAK, J.; PINKIEWICZ, H.

Studies on the pathogenesis of experimental influenza in white mice.
Med. dosw. mikrob. 10 no. 1:15-31 1958.

1. Z Katedry Mikrobiologii Lekarskiej A. M. w Lublinie Kierownik:
prof. dr J. Parnas.

(INFLUENZA, exper.

comparative pathogen. in mice after infect. with various
strains & by various routes (Pol))

TURCZSKI, Janusz; FAMIA SKI, Jerzy; PIWIL, Iwo

Model studies on scattering losses in transformers. Elektryka
Lodz no.14:05-11 '63.

1. Katedra Maszyn Elektrycznych i Transformatorow, Politechnika,
Lodz.

PINKIN, A. N.

Remont chasov (Repairing of clocks and watches) 3. Izd. i ispr. izd. sovrem., 1952.
210 p. illus., tables. "Literatura": p. (17")

SD: N/5
744.5
.17
12.2

HARVEY, A. M.

Recurrents, Inc., 1000, 10th Street, N.W., Washington, D.C.
Telephone: 270-1111

Proposed to be used in connection with:

1.20: 7-10-76

2. Manifestations of Mechanized Engineering in the Development of
Germany, 1945.

PINKIN, ABRAT MIKHAYLOVICH

N/5
744.5
.P6

Remont chasov (Repairing of clocks and watches) Moskva, KOIZ, 1952-
v. illus., diafrs., tables.

Includes bibliography.

Lib. has: 1952
1957

PINKINA, L. N.

"Studies of polycondensation reactions. III. On the influence of the components ration upon the growth of the polyamide chain." Rafikov, S. R., Korshak, V. V. and Pinkina, L. N. (p. 1003)

SO: Journal of General Chemistry (Zhurnal Obshchei Khimii) 1941, Volume 11, no. 9-10.

FINKINA, L. N.

USSR/Chemistry - Acetylene, Derivatives
Chemistry - Olefins, Hydration of

App. 1

"Acetylene Derivatives: No 66, Mechanism of the Hydration and Cyclization of Dienes," I. N. Nazarov, L. N. Finkina, Inst Org Chem, Acad Sci U.S.S.R, 54 pp

"Zhur Obshch Khim" Vol XVIII (LXXX), No 4

4-Methyl-1-vinylethynyl- Δ^1 -cyclohexene, on heating in methanol in the presence of sulfuric acid and mercury sulfate, is hydrated to allyl-4-methyl- Δ^1 -cyclohexenylketone. The latter compound, under the influence of phosphoric or hydrochloric acid, is readily cyclized into 1,6-dimethyl-4,5,6,7-tetrahydroindan, which is also formed when dienin is heated with phosphoric acid. Submitted 7 Apr 1967.

PA 8/49 Th1

FINKINA, L. N.

USSR/Chemistry - Acetylene, Derivatives
Chemistry - Olefins, Hydration of

Apr 11

"Acetylene Derivatives: No 17, Mechanism of the Hydration and Cyclization of Dienes," I. N. Nazarov, L. N. Finkina, Inst Org Chem, Acad Sci USSR, 54 pp

"Zhur Obshch Khim" Vol XVIII (LXXX), No 4

5-Methyl-1-vinylethynyl-¹-cyclonexene, on heating in methanol solution in the presence of sulfuric acid and mercury sulfate is hydrated to allyl-5-methyl-¹-cyclohexenylketone, which, under the influence of phosphoric acid is cyclized to 1, 5-dimethyl-4, 5, 6, 7-tetrahydroindan-3-one. The latter can also be prepared directly by heating diene with phosphoric acid at 60 - 65°. It can be transformed into 1, 5-dimethylindan by heating with phosphoric acid at 170 - 190°. Submitted 7 Apr 1947.

PA 8/49 T42

PINKIN², I. N.

"Acetylene Derivatives. 102. Mechanism of the Hydration and Cyclization of Diacines. XXI. Hydration of 5-Isopropyl-6-methyl-1,5-pentadien-2-one into 5-Isopropyl-(methyl-1,5-hexadien-4-one and its Cyclization into 1-Isopropyl-2,7,3-trimethyl-hepta-2,4-cyclononen-5-one. A New Instance of the Cyclization of Substituted Vinylallylketones which do not have a Free Hydrogen in the Vinyl Radical."

Nazarov, I. N. and Pinkin, L. N. (Inst Org Chem, Acad Sci USSR) (P. 1270)

SO: Journal of General Chemistry (Zhurnal Obshchei Khimii) 1949, Vol. XIX, N^o. 10

CA

Acetylene derivatives CXX. Mechanism of hydration and cyclization of diynes. 27. Transformation of 2-methyl-1-vinylcyclohexanone into 1,2-dimethyl-3-oxahydroinden-3-one. A new method of synthesis of polycyclic ketones containing a cyclopentanone ring with angular substituents. I. N. Nazarov and L. N. Pankau. *Vser. Osnovatel. Akad. Nauk. Gen. Chem.* 20, 2039 (2a) (1950). *J. Am. Chem. Soc.* 72, 2686 (1950). Adding with cooling 73 g 2-methyl-

cyclohexanone and 50 ml CH_2Cl_2 $\text{CH}_2=\text{CH}$ in Et_2O to 10 ml NaOH and 17 g pyridine-KOH and letting stand overnight gave 91% 2-methyl-1-vinylcyclohexanone (I). In the latter with $\text{Br}-\text{NaBH}_4$ gave 2-methylcyclohexanone and, in 102° authentic specimen made from citraconic anhydride and butadiene by heating 5.5 hrs at 102°, yielding the anhydride, b.p. 108–111°/4–10 mm. Stirring 40 hr II with 270 g 93% MeOH , 0.4 ml HgSO_4 , and 2 g HgSO_4 2.5 hrs at 60° with gradual addition of 2 g HgSO_4 gave 46% 2,6-dimethyl-3-cyclohexen-1-yl ketone (IV), b.p. 65–71°/1 mm, d₄²⁰ 0.8531, hydrogenated over Pt oxide to P_t 2.1 mm², d₄²⁰ 0.8531, hydrogenated over Pt oxide to P_t 2.1 mm², d₄²⁰ 0.8531, hydrogenated over PtO_2 to P_t 2.1 mm², d₄²⁰ 0.8531, b.p. 67°/1.4 mm, d₄²⁰ 0.8531, heat alone in 102°/3° from MeOH , hydrogenation is best done in AcOH , as in IV only 2-H add readily. Chromolysis of IV yields 11C-CH₂-AcOH and $\text{Me}_2\text{NaC}_6\text{H}_5\text{CH}_2\text{CO}_2\text{Na}$, which on oxidation by $\text{Br}-\text{NaBH}_4$ and $\text{Me}_2\text{NaC}_6\text{H}_5\text{CH}_2\text{CO}_2\text{Na}$ stirring 11.9 g IV with 12 ml HgSO_4 and 1.76 g HgSO_4 at room temp and 6 hrs at 60° gave 16.6 g III, but which described above, and another, in 102°/3°, although two 2,6-dimethylphenylketones form, only one, in 125°/3° was isolated in the pure state, 1.16 g, with b.p. 114–116°/1–170–210 mm, at room temp, and 8 hrs at 60°/3° gave Me_2Na in MeOH gave 1.2 g product 18.3%, apparently also identical with III, giving predominantly the second isomer, in 120°, and 2,3-dimethylcyclohexanone in 18%.

The acid analog failed to change on heating with NaBH_4 . **CXXI. Mechanism of hydration and cyclization of diynes. 28. Hydration of 2,6-dimethyl-1-vinyl-2-ethyl-2,6-dimethylcyclohexen-1-yl ketone in cyclization**

10

ca

reactions. I. N. Novikov and M. S. Burmistrova. *Ibid.* 27(1), 8. To 80 g. iron KOW, 250 ml. H_2O , and 25 g. CH_3COCl was added, with ice cooling, in 1 hr. 100 g. 2,4-dimethylheptanone and 10 g. $\text{CH}_3\text{COCH}_2\text{CH}_2$ and 100 ml. H_2O , letting stand 1 hr. at -5° and overnight at room temp. gave 115 g. 2,4-dimethylheptane which contained 16.1% allyl. Hydrogenation over Pt in H_2O to the 1,6-analog, $\text{mp} 21^\circ$, $d_2^{20} 0.850$, over Pt/HgSO_4 in the same way, gave 115 g. hydrogenated oil ($d_2^{20} 0.855$), while stirring the unsatd. oil ($d_2^{20} 0.850$) with 100 ml. 10% HgSO_4 with a little pyrogallol 4 hrs. at 0° gave 115 g. 2,4-dimethylhexanone with a yield of 78%. In the same way, 1,6-dimethyl-1-hexylcyclohexene (I), $\text{mp} 78^\circ$, was allylated with 10% HgSO_4 hydrogenated over Pt to the 1,6-analog, 2,6-dimethyl-1-hexylcyclohexene, $\text{mp} 86-87^\circ$, and 1,6-dimethyl-1-hexylcyclohexene, $\text{mp} 105^\circ$ (from MeOH and HgSO_4), stirring 115 g. 10% HgSO_4 and 100 ml. H_2O gave 20.5 g. allyl 2,6-dimethylcyclohexene (II), $\text{mp} 45^\circ$, $d_2^{20} 0.857$. Repeating with 425 g. 10% MeOH , 2.6 g. HgSO_4 , and 100 ml. H_2O gave 110 g. most of the above ketone with the 2-methylcyclohexene, heating the mixt. with 5% $\text{MeOH}/\text{HgSO}_4$ 2 min. at 10° , 50°, and 100 mm. gave 115 g. II. In CH_2Cl_2 and HgSO_4 , hydrogenation of this gave the monohyd. oil, $\text{mp} 22^\circ$, $d_2^{20} 0.850$, $d_2^{20} 0.857$, while allylation gave 115 g. II and 1,6-dimethyl-1-hexylcyclohexene, $\text{mp} 120-125^\circ$, yielding the same ketone, $\text{mp} 121-122^\circ$. Stirring 115 g. II with 115 ml. 10% HgSO_4 at 0° for 22 hrs. at room temp. gave 45 g. of the compd. (III), $\text{mp} 92.5-93^\circ$, $d_2^{20} 0.902$, $d_2^{20} 0.927$. Some ketone, $\text{mp} 108-109^\circ$ (from MeOH), the same peroxide is obtained in better yield by hydration with 10% HgSO_4 in the presence of HgSO_4 after 24 hrs. at 0° .

The peroxide does not take up H with a Pt catalyst in AcOH .
The stronger HgSO_4 , gives best yields.



*A**Organic chemistry - 10*

Acetylenes derivatives. CXXI. Mechanism of hydration and cyclization of diynes. 27. Transformation of 7-methyl-1-(vinylethylnyl)cyclohexanol into 3,3a-dimethyl-3a,4,5,6,7,7a-hexamethylbenzene. A new method of synthesis of polycyclic ketones containing a cyclopentimone ring with angular substituents. I. N. Nazarov and L. N. Punkina. *J. Gen. Chem. U.S.S.R.* 20, 2079-91 (1950) (Engl. translation).—See *C.A.* 43, 80306. CXXII. Mechanism of hydration and cyclization of diynes. 28. Hydration of 2,4-dimethyl-1-(vinylethylnyl)cyclopentane and the behavior of the thus formed allyl 2,4-dimethyl-1-cyclopenten-1-yl ketone in cyclization reactions. I. N. Nazarov and M. S. Burmistrova. *Ibid.* 2091-7.—See *C.A.* 43, 60371. B. L. M.

BUKKINA, L. N.

BUKKINA, L. N. -- "Hydration of Vinyl-delta-prime-Cyclohexenylacetyles and the Cyclization of Allyl-delta-prime-Cyclohexenylketones. A New Type of Cyclization of Substituted Vinylallylketones without a Free Hydrogen in the Vinyl Radical." Sub 30 Sep '62, Inst of Organic Chemistry, Acad Sci USSR. (Dissertation for the Degree of Candidate in Chemical Sciences).

TO: Vechernaya Moskva January-December 1962

62-58-3-1/10

AUTHORS: Knunyants, I. L. , Sterlin, R. N. , Pinkina, L. N.,
Dyatkin, B. L.

TITLE: Reactions of Fluorolefins (Reaktsii ftorolefinov)
Communication 7. Addition Compounds of Acid Chlorides to
Vinylidene Fluoride and Trifluoroethylene (Sooobshcheniye 7.
Prisoyedineniye khlorangidridov kislot k ftoristomu vinili-
denu i trifloretilenu)

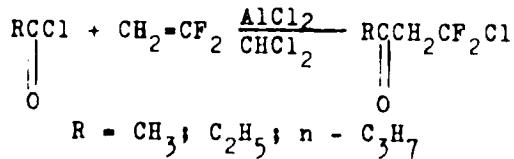
PERIODICAL: Izvestiya Akademii Nauk SSSR, Otdeleniye Khimicheskikh Nauk
1958, Nr 3, pp. 296 - 299 (USSR)

ABSTRACT: The addition of alkyl halides discovered by Kondakov was later developed by others. In the present paper the authors show that such fluorolefins as vinylidene fluoride and trifluoroethylene (in the presence of nonaqueous AlCl_3) possess the capability of combining with carboxyl chlorides and thereby forming the corresponding fluoro-substituted ketones. Vinylidene fluoride very readily combines with the acid chlorides of butyric acid and propionic acid at a temperature of -5 to -10°C in the presence of equivalent quanta of AlCl_3

Card 1/2

62-58-3-6/30
Reactions of Fluorolefins. Communication 7. Addition Compounds of Acid Chlorides to Vinylidene Fluoride and Trifluoroethylene

pure chloroform, where alkyl-2-chloro-2,2-difluoroethylketones with yields of 44,48 and 33 % form:



At the same time substances form which correspond to the products of the partial or complete substitution of fluorine in chlorine and the products of further condensation. The authors obtained: methyl-2-chloro-2,2-difluoroethylketone and methyl-2-chloro-1,2,2-trifluoroethylketone. There are 8 references, 2 of which are Soviet.

SUBMITTED: November 3, 1956

Card 2/2

5(3)
AUTHORS:Knunyants, I. L., Sterlin, R. N.,
Yatsenko, R. D., Pinkina, L. N.

SOV/62-58-11-11/26

TITLE:

Reactions of Fluoro Olefins (Reaktsii ftorolefinov)
Communication VIII. Reactions of Perfluoro Vinyl Magnesium
Halides (Soobshcheniye 8. Reaktsii
perftorvinilmagniygalogenidov)

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,
1958, Nr 11, pp 1345-1347 (USSR)

ABSTRACT:

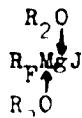
In the present paper the authors demonstrated that by the activation of magnesium with ethyl bromide and by carrying out the reaction in ester at -30 to -20° a practically quantitative consumption of magnesium can be achieved. By the decomposition of the reaction mass with diluted sulfuric acid 70 % of trifluoro ethylene could be separated. It was demonstrated that under the mentioned conditions perfluoro vinyl bromide and perfluoro vinyl chloride do not react with magnesium and that they are unchanged after the end of the reaction. An organomagnesium compound $\text{CF}_2=\text{CFMgBr}$ in a yield of up to 45 % could be formed from perfluoro vinyl bromide in tetrahydro furan. In this case it was not even necessary to

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Reactions of Fluoro Olefins.
Communication VIII. Reactions of Perfluoro Vinyl
Magnesium Halides

SCV/60-5E-11-11/26

activate magnesium with ethyl bromide. Apparently the assertion that an intensification of the basicity of the solvent favors the formation of $R_F MgJ$ on the basis of its stabilization in the form of a complex of the



type, is justified. As the result of the processing of $CF_2 - CFMgJ$ with solid carbon dioxide in ester solution at -40° and the subsequent decomposition of the reaction mass with 2N sulfuric acid solution perfluoro acrylic acid was obtained in a yield of 40 %. Henne (Ref 6) formerly obtained this acid by a complex and very slow method. The found method can be recommended without doubt for preparation.

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By processing the ester solution of the perfluoro acrylic

Reactions of Fluoro Olefins.

SOV/62-58-11-11725

Communication VIII. Reactions of Perfluoro Vinyl
Magnesium Halides

acid with a calculated amount of diazomethane the methyl
ester of perfluoro acrylic acid was obtained. There are
8 references, 1 of which is Soviet.

SUBMITTED: March 4, 1957

Card 3/3

5 (3)

AUTHORS:

Sterlin, R. N., Znunyants, I. L.,
Pinkina, L. N., Yatsenko, R. D.

SOV/62-59-8-29/42

TITLE:

Tetraperfluorovinylsilane

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,
1959, Nr 8, pp 1492-1493 (USSR)

ABSTRACT:

Starting from a consideration of the reaction of tetrachloro-silicon with alkyl- and aryl silanes and other organic silicon (or magnesia) halogenides, the present paper describes the attempted gradual substitution for the Cl-atom in SiCl₄ of a perfluorovinyl group. As expected, the introduction of such a group caused a decrease in the electron density in the central Si-atom. Thus the substitution of further groups is progressively facilitated. The tetrafluorovinylsilane is stable in aqueous acid solutions; in bases it is quantitatively split into trifluoroethylene which has been identified by its dibromide. The reaction is described in the experimental part. There is 1 reference.

SUBMITTED: February 11, 1959

Card 1/1

100
20043-100000

AUTHORS: Sternin, R. N., Flukina, Taiserk, R. D., Kurnyanov, I. L.

TITLE: ~~ANALYSIS OF POLY(1,1-DIFLUOROVINYLIC ARYL DERIVATIVES)~~
and Sci

PERIODICAL: Khim. i Tekhnologiya Polymernykh Materialov, No. 1,
1977, p. 10-13 (USSR)

ABSTRACT: The authors report the synthesis of a series of poly(1,1-difluorovinylidene aryl) polymers. These polymers are obtained by the action of vinylidene dichloride on the corresponding aryl compounds in the presence of aluminum chloride. The authors also report the synthesis of tri-(trifluoromethyl)-1,1-difluorovinylidene arylarsazine, which was obtained from perfluorovinylmagnesium bromide and a arsenide. Perfluorovinylidenearsazine (a new product) was obtained by decomposition of 10-perfluorovinyl-*n*,10-di-vinylarsazine with liquid HCl. The obtained substances have the following properties: tri-(trifluoromethyl)-

Card 1 of

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5

Report of the Joint Planning Board
Detailed Summary Report

The following report is a detailed summary of the activities of the Joint Planning Board during the period from January 1, 1986 through December 31, 1986. It includes a description of the organization and functions of the Board, its major accomplishments, and its recommendations for future action.

The Joint Planning Board was established by the Director of Central Intelligence on January 1, 1986, to provide a forum for the joint planning of intelligence activities. The Board consists of representatives from all intelligence agencies, including the Central Intelligence Agency, the Federal Bureau of Investigation, the National Security Agency, the Defense Intelligence Agency, and the National Security Council. The Board's primary function is to coordinate intelligence activities and to develop joint intelligence requirements. It also provides advice and guidance to the Director of Central Intelligence on intelligence matters.

The Board has made significant contributions to the development of joint intelligence requirements and to the coordination of intelligence activities. It has helped to ensure that intelligence resources are used effectively and efficiently. The Board has also provided valuable advice and guidance to the Director of Central Intelligence on a variety of intelligence issues.

SUBMITTED: [Signature]

Chairman

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5"

AUTHORS:

S. V. K. R. S., P. G. G., L. N., K. S. S., J. S.

S. C. I. T. - 1

TITLE:

Radical Exchange in Perfluorobromoethane and Diphenyl

PERIODICAL:

Khimicheskaya i fizicheskaya chistoostroenie i ogranichenie
kremniyazernogo perekhoda, 1970, No. 1, p. 10-13.

ABSTRACT:

Radical exchange was studied in the system perfluorobromoethane-phenyl iodide-phenylmagnesium bromide, phenyl iodide-arsenic trioxide and phenyl iodide-(triphenylphosphine)-arsine (1.1% each), phenyl iodide, phenyl iodide (0.8%), phenyl iodide-phenylmagnesium bromide, phenyl iodide-arsenic trioxide and phenyl iodide-(triphenylphosphine)-arsine (1.1% each), phenyl iodide, phenyl iodide (0.8%), phenyl iodide-phenylmagnesium bromide, phenyl iodide-arsenic trioxide and phenyl iodide-(triphenylphosphine)-arsine (1.1% each). The following radical exchange reactions were obtained: $R'X \rightarrow R'M-X + RX$ radical exchange takes place when R' and R are sufficiently different in polarity. The authors are: S. V. K. R. S., P. G. G., L. N., K. S. S., J. S. The U.S. performers are: C. R. P. P.,

Radius of the Earth = R_E = 6.37×10^6 m
Radius of the Sun = R_S = 6.96×10^8 m

GOV'T OF THE U.S.

L. F. Mccormick, E. T. McPherson, J. A. McPhee, G. W. McRae,
C. C. McRae, H. H. McRae, R. L. McRae, J. A. McRae,
(Continued on page 14) P. Thompson, D. A. Warner, T. J. Warner,
R. C. Warner, J. O'Connor, J. O'Connor, J. O'Connor

SUBMITTED: A

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"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001340920007-5

ATTACHMENT

TITLE:

PERIODICITY:

CLASSIFICATION:

DATE: 19



CLASSIFICATION:

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Thus, we have shown that $\mathcal{D} = \mathcal{D}'$. This completes the proof.

Figure 1. The relationship between the number of species and the area of forest cover.

$$(\text{IP} - \text{IPF})_{\text{obs}} + \text{IPF}_{\text{pred}} \rightarrow (\text{IP} - \text{IPF})_{\text{pred}}$$

EDUCATIONAL INSTITUTIONS

Cann. 44

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53630

1287, 2209, 1266

S/362/60/300/100/100/100
BC13/3076

AUTHORS: Sterlin, R. N., Yatsenko, R. D., Pinkina, L. N.,
Knunyants, I. I.

TITLE: Perfluoro Derivatives of Nonmetals

PERIODICAL: Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh
nauk, 1960, No. 11, pp. 1991 - 1997

TEXT: The preparation of perfluoro derivatives of phosphorus, arsenic,
and antimony is described. From the reaction of perfluorovinyl magnesium
iodide with AsCl_3 , PCl_3 , and SbCl_3 in ether solution only tertiary de-
rivatives were obtained: tri-(trifluorovinyl)arsine, tri-(trifluoro-
vinyl)phosphine, and tri-(trifluorovinyl)stibine. Primary and secondary
derivatives were not formed in this process. Perfluorovinyl dichloro-
arsine was obtained by splitting 10-alkyl-5,10-dihydrophenarsazine with
dry HCl (Ref.4). A corresponding perfluorovinyl derivative was obtained
in a quantitative yield as a result of the reaction of perfluorovinyl
magnesium iodide with adamsite. Perfluorovinyl chloroarsine was

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Perfluoro Derivatives of Nonmetals

S/062/60/000/S11/006/S16
BC13/EC78

isolated in a practically quantitative yield by the reaction of $\text{CF}_2\text{-CFAs}(\text{C}_6\text{H}_4)_2\text{NH}$ with liquid HCl. By treating the tetraethyldiamide of phosphorous acid chloride and the tetraethyldiamide of ethyl phosphinic acid with dry, gaseous HCl in xylol solution, phosphorus trichloride, and ethyldichlorophosphine, respectively, were obtained. From the reaction of perfluorovinyl magnesium iodide with the tetraethyldiamide of phosphorous acid chloride, the tetraethyldiamide of perfluorovinyl phosphinic acid was obtained. This was converted into trifluorovinyl dichlorophosphine by reaction with dry HCl in ether solution. By treating the latter with antimony trifluoride, perfluorovinyl difluorophosphine was obtained. In a similar manner, the diethylamide of di-(trifluorovinyl) phosphinic acid was obtained from $(\text{C}_2\text{H}_5)_2\text{NPCl}_2$ and perfluorovinyl magnesium iodide. By decomposing it with dry HCl, di-(trifluorovinyl)chlorophosphine was synthesized. By treating the latter with antimony trifluoride, di-(trifluorovinyl)fluorophosphine was obtained. As opposed to the trifluoromethyl derivatives of arsenic and phosphorus, the prepared tri-(trifluorovinyl)arsine and tri-(trifluorovinyl)phosphine do not

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86479

Perfluoro Derivatives of Methacrylic Acid

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separate trifluoroethylene when heated. Thus, the perfluorovinyl derivative in the said compounds does not show any properties of pseudohalogens. Ye. P. Shcherbina and L. F. Razgovorov assisted in this work. There are 8 references: 2 Soviet.

SUBMITTED: June 4, 1980

Card 3/3

KNUNYANTS, I.L.; STERLIN, R.N.; TYULENEVA, V.V.; PINKINA, L.N.

Pseudohalide properties of perfluoroalkenyl radicals in esters of
perfluoroalkenylphosphinic acids. Izv. AN SSSR. Otd.khim.nauk
no.6:1123-1127 Je '63. (MIRA 16:7)

1. Institut elementoorganicheskikh soyedineniy AN SSSR.
(Phosphinic acid) (Radicals (Chemistry))

J. 12718-63EPR/EPP(c)/EMP(1)/EWI(n)/BIS ASD Po-4/Po-4/Pr-4 RP/MAY/WW
ACCESSION NR: AP3002298 S/0062/63/006/1123/1127 72AUTHOR: Kunyants, I. L.; Sterlin, R. N.; Tyuleneva, V. V.; Pinkina, L. N. 71TITLE: Pseudohalo properties of perfluoroalkenyl radicals in esters of perfluoro-
alkenylphosphoric acid

SOURCE: AN SSSR. Izv. Otdeleniya khimicheskikh nauk, no. 6, 1963, 1123-1127

TOPIC TAGS: perfluorovinyl phosphoric acid, perfluoropropenyl phosphoric acid,
perfluoroisobutetyl phosphoric acid, perfluoroisobutetyl phosphoric acid, methyl
perfluoropropyl phosphoric acid, hydrolysis, diisopropyl ester

ABSTRACT: The following new esters of perfluoroalkenylphosphoric acids were prepared; chemical and physical data is given: diethyl ester of 1,2-difluoro-2-chlorovinyl phosphoric acid; diisopropyl ester of perfluorovinyl phosphoric acid; diisopropyl ester of perfluoropropenyl phosphoric acid; diisopropyl ester of perfluoroisobutetyl phosphoric acid; diisobutyl ester of perfluoroisobutetyl phosphoric acid; and isopropyl ester of methyl perfluoropropenyl phosphoric acid. The perfluoropropenyl and perfluoroisobutetyl radicals at the pentavalent P show pseudohalo properties: alkaline hydrolysis of the above esters gives the corresponding fluoroalkenes, alcohols and alkali phosphate; the fluorovinyl phosphoric

Card 1/2

L 12718-63

ACCESSION NR: AP3002298

acids did not give trifluoroethylene on hydrolysis. Orig. art. has: 3 tables, 4 formulas and 2 figures.

ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR
(Institute of Organoelemental Compounds, Academy of Sciences SSSR)

SUBMITTED: 15Jan63

DATE ACQ.: 16Jul63

ENCL: 00

SUB CODE: 00

NO REF Sov: 004

OTHER: 001

Card 2/2

PIN'KOVSKIY, Gleb Stanislavovich; SARSER, A.I., retsenzent;
SLAVOROSOV, A.Kh., red.izd-va; ZHIVRINA, G.V., tekhn.
red.; ROMANOVA, N.V., tekhn. red.

[Surveying operations during mine shaft sinking] Marksheiderskie raboty pri sooruzhenii shakhtnykh stvolov. Moskva,
Izd-vo "Nedra," 1964. 150 p. (MIRA 17:3)

VAGIN, Gennadiy Ivanovich; PIN'KOVSKIY, Gleb Stanislavovich; CHERNEGOVA,
E.N., red. izd-va; SHKLYAR, S.Ya., tekhn. red.

[Using portable formwork in supporting mine shafts] Kreplenie
stvolov shakht s primeniem perevizhnykh opalubok. Moskva,
Gosgortekhizdat, 1962. 146 p. (MIRA 15:11)
(Mine timbering)
(Concrete construction--Formwork)

PIN'KOVSKIY, G.S., inzh.; ABLETS, V.I., inzh.

Destructive action of corrosive waters on concrete shaft supports in the Krivoy Rog Basin. Shakt. strol. 6 no. 7:e-1
Jl '62. (Jul 1962)

1. Treat krivbasus maki-to; rokhodka.
(Mine water)
(Krivoy Rog Bas n-- Concrete construction,

PIN'KOVSKIY, G.S., inzh.

Using length-tapping devices for depth measurement in shaft
sinking. Shakht.stroi. 6 no.2:18-23 F '62. (MIRA 15:2)

1. Trest Krivbassshakhtoprokhodka.
(Shaft sinking)(Measuring tapes)

VAGIN, G.I. , inzh., ~~RIL'INOVSKIY~~, G.S.

Borehole charge firing in stages during shaft sinking. Shakh.:.
stroj. no.6:26-27 Ja '59. (MIRA 12:0)

1. Trust Krivbasshakhtoprokhodka.
(Shaft sinking) (Blasting)

PIN'KOVSKII, S.I.

Types of river channels in central and southern Siberia. Trud GGI
no..94:87-114 '62. (MIRA 15:7)
(Siberia--Rivers)